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## **ANSWERS**

By looking at the periodic table (section 6 of the IB data booklet), it can be seen that the element is lithium. 30 For CdTe, percentage by mass =  $112.41 \times 100 = 46.84\%$   $112.41 + 127.60$  Overall, CdS has the highest percentage by mass of cadmium. You could also approach this question by considering the Ar of the other element in the compound.

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