

## Chapter 12 Stoichiometry Vocabulary Review Answer Key

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Chapter 12 Review Stoichiometry Important Vocabulary • Stoichiometry- The calculations of quantities in chemical reactions in a subject of chemistry • Mole ratio- a conversion factor derived from the coefficients of a balanced chemical equation interpreted in terms of moles

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Chapter 12 REVIEW: Stoichiometry, Theoretical, Actual & Percent yield Part I. Stoichiometry 1.  $1 \text{ ZnI}_2 \rightarrow 1 \text{ Zn} + 1 \text{ I}_2$  How many grams of iodine will you produce if you begin your rxn with 56.7g of zinc iodide?  $56.7 \text{ g} \cdot 1 \text{ mole} \cdot 1 \text{ mole} \cdot 253.8 \text{ g} \cdot 319.19 \text{ g} \cdot 1 \text{ mole} \cdot 1 \text{ mole} = 45.08 \text{ grams of Iodine}$  2.  $1 \text{ Pb}_2(\text{CO}_3)_3 + 6 \text{ K} \rightarrow 2 \text{ Pb} + 3 \text{ K}_2\text{CO}_3$

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STOICHIOMETRY 12 Vocabulary Review Match the correct vocabulary term to each numbered statement. Write the letter Of the correct term on. the

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line. Column A 1. the starting materials in a chemical reaction 2. a conversion factor derived from the coefficients of a balanced chemical equation interpreted in terms Of moles

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Chapter 12 Stoichiometry 289 ... Section Review Objectives ... Vocabulary • stoichiometry Part A Completion Use this completion exercise to check your understanding of the concepts and terms that are introduced in this section. Each blank can be completed with a term, short ...

### **05 CTR ch12 7/9/04 3:34 PM Page 289 THE ARITHMETIC OF ...**

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