

Chemistry Midterm Study Guide

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(look on study guide) -The atomic number of this atom is 2, based on the number of protons -The mass number for this atom is the sum of the protons and electrons -This is a +1 charged ion because it has fewer electrons than when it's neutral

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A graduated cylinder contains 44.2 milliliters of water. a 48.6-g piece of metal is carefully dropped into the cylinder. when the middle is completely covered with water the water rises to the 51.3 milliliter mark. what is the density of the metal? 6.8 g/mL

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Chemistry 1027b Midterm Study Guide Matter Properties of matter; physical and chemical (what is the difference?) A physical property is one that can be observed without any change in composition. A chemical property is one that is observable in chemical reactions and changes the identity of one or more reactants.

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Chemistry Midterm Study Guide. Just from \$13,9/Page. Get custom paper. 0.2×10^{23} , Mass number (Z) is the sum of protons plus neutrons •Isotopes of an element have the same number of protons, but different numbers of neutrons. •Isotopes of an element have the same number of protons, but different numbers of neutrons.

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Midterm Exam 2 Study Guide: CHEM-3-94926 - Introductory Inorganic Chemistry Midterm Exam 2 Study Guide The second midterm exam will cover Lectures 8-13 & Chapters 6-8 in the textbook. See below for a comprehensive summary of the topics included on the exam.

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