

Chemistry Practical Viva Questions With Answers

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Chemistry Practical Viva Questions With

Viva Questions with answers. Question.1.What is a standard solution ? Answer. A solution whose strength is known is called a standard solution. Question.2.What is a normal solution ? Answer. A solution containing one gram-equivalent mass of the solute per litre of the solution is called a normal solution. Question.3.What is the equivalent mass of KMnO_4 when it acts as oxidizing agent in acidic medium ? Answer.

Chemistry Practical Class 12 Volumetric Analysis Viva ...

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The examiner may ask questions like "What if this function would have been replaced by some other function?", "Explain how this function will work line by line" or "How can we implement this statement in another way?" etc. So make them perfect.

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Chemistry Practical Viva Questions With Answers

This collection of chemistry test questions is grouped according to subject. Each exam has answers supplied at the end. They provide a useful study tool for students. For instructors, they are a good resource for homework, quiz, or test questions, or practice for the AP Chemistry test. 01.

Quiz Yourself Using These 20 Practice Chemistry Tests

Practical Viva Questions Organic Chemistry Interview Questions Answers, Practical Viva Questions Placement Papers, Practical Viva Questions Technical, HR Interview Questions, Practical Viva Questions Aptitude Test Questions, Practical Viva Questions Campus Placements Exam Questions in categories , C, C Sharp, VB.NET, SQL PLSQL, Core Java, HR Questions, Organic Chemistry, Lab Technicians, Civil ...

Practical Viva Questions Organic Chemistry Interview ...

To prepare a standard solution and determine the strength of a given solution of hydrochloric acid by titrating it against the standard solution prepared.

Quantitative Estimation (Viva Voce) : Class 11 : Chemistry ...

Viva Questions with Answers on Quantitative Estimation (Volumetric Analysis) Chemistry Lab Manual NCERT Solutions Class 11 Chemistry Sample Papers. 1. Why should weights not be lifted with hand ? Ans. This causes error in the weighing because some matter may be transferred from the hand to the weight. 2.

Viva Questions with Answers on Quantitative Estimation ...

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Chemistry Practical Viva Questions With Answers

Chemistry practical viva questions and answers are discussed, which may be helpful for school and college students.

CHEMISTRY PRACTICAL VIVA QUESTIONS AND ANSWERS

I am giving here some sample questions for chemistry viva - The reason of heating done in Oxalic Acid ,KMnO₄ titration? Why is H₂SO₄ used in the replacement of HCl? Briefly discuss the principle of salt analysis ? Give some of the examples of salt with Colours? Discuss the significance of flame test . Why we use HCl in a Flame Test?

Chemistry viva questions for practicals class 11 - Brainly.in

CHEMISTRY LAB VIVA Questions. VIVA QUESTIONS - PRACTICALS 1. DETERMINATION OF HARDNESS OF WATER. 1. What is temporary hardness?

CHEMISTRY LAB VIVA Questions by ramesh kumar - Issuu

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Chemistry Practical Viva 3

Viva Voce – Questions on Titration Practicals. Spread the love. Why is diluted sulphuric acid [H₂SO₄] added while preparing standard Mohr's solution? It is added to prevent hydrolysis of ferrous salt. Can you use hydrochloric acid [HCl] or nitric acid [HNO₃] in potassium permanganate [KMnO₄] titrations? Why?

Class XII CBSE Chemistry Practicals - Viva Questions on ...

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This is a collection of 10 chemistry test questions with answers dealing with unit conversions. Question 1 . Convert the following measurements into m. a. 280 cm b. 56100 mm c. 3.7 km Question 2 . Convert the following measurements into mL. a. 0.75 liters b. 3.2 x 10⁴ μL c. 0.5 m³.

Unit Conversions Test Questions - ThoughtCo

Organic Chemistry Questions The Covalent Bond 1. The hybridization of the central carbon in CH₃C≡N and the bond angle CCN are a. sp², 180°. b. sp, 180°. c. sp², 120°. d. sp³, 109°. 2. Which of the following statements about an sp hybridized carbon is FALSE? a. It is divalent. b. It forms bonds that are linear. c. It has two p orbitals. d.

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