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eraser, and scientific calculator to
complete the test. You will be given a
periodic table. Useful formulas and the
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the largest energy change per mole of fluorine? A. $F_2(g) \rightarrow F_2(l)$ B. $19.9 F + 4$
 $2 He \rightarrow 1.0 n + 22.11 Na$ C. $2 F_2(g) + 2$
 $H_2O(l) \rightarrow O_2(g) + 4 HF(aq)$

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(Seniors _____) REMINDER: DON'T FORGET YOUR NON-GRAPHING SCIENTIFIC CALCULATOR THE DAY OF THE EXAM! You will be provided with a periodic table with the polyatomic ion chart and equations that appear on the last page of this packet.

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EXAM STUDY GUIDE

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Practice Exam 1. Answers to PE1.

Practice Exam 2. Answers to PE2.

Practice Exam 3. Answers to PE3 pgs1-5.

CH141 Practice Exam III Key B. Practice
Final Exam Problems. PF answers pg1-6.

CH141 Practice Final Key II (pages 6-12)

CH141 Exam I 2016 with Answers.

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Introduction to Chemistry Final

Examination Time allowed: 09:00 -

18:00 (9 Hours) This examination paper
contains two sections: Section A (2 short
answer questions, each worth 10 marks

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for a total of 20 marks) and Section B (a descriptive summary, worth 15 marks). The maximum mark available is 35. There are a total of 3 pages including this cover page.

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