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## **Gas Laws And Gas Stiochiometry**

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Gas stoichiometry. Gas stoichiometry is the quantitative relationship (ratio) between reactants and products in a chemical reaction with reactions that produce gases. Gas stoichiometry applies when the gases produced are assumed to be ideal, and the temperature, pressure, and volume of the gases are all known. The ideal gas

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law is used for ...

## **Stoichiometry - Wikipedia**

Find the volume of 1 g of H<sub>2</sub> gas in litres at N.T.P b.Find the volume of 20g H<sub>2</sub> at NTP. c.What is the volume occupied by  $6.022 \times 10^{23}$  molecules of any gas at NTP?

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## **Mole Concepts Numericals with Detailed Solutions**

A 1.65 gm sample of  $\text{FeS}$ , was oxidized by excess oxygen & the products were  $\text{SO}_2$  and  $\text{Fe}^{2+}$ , The  $\text{SO}_2$  gas produced was passed through an acidified solution of 40ml  $\text{Ba}(\text{MnO}_4)_2$  produced  $\text{SO}_4^{2-}$  &  $\text{Mn}^{2+}$ .

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