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Review Module /  
Chapters 25-28 17 In  
your notebook, solve  
the following problems.

SECTION 25.1

HYDROCARBONS 1.

Name this hydrocarbon  
according to the IUPAC  
system. 2. Name this

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hydrocarbon according to the IUPAC system. 3.

Write the structural formula for each of the following compounds.

a. 3-ethyl-2,3-dimethyl pentane

b. 3,4-diethylhexane

## **25 Hydrocarbon Compounds Practice Problems**

[DOC] Chp 22

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HYDROCARBONS 22.1

Section Review

Objectives e Describe

the relationship

between number of

valence electrons and

bonding in carbon

Define and describe

alkanes Relate the

polarity of

hydrocarbons to their

solubility substituent

alkyl group branched-

chain alkane

Vocabulary

hydrocarbons alkanes



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straight-chain alkanes

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**UHS Science**

Modern Chemistry 181

Organic Chemistry

CHAPTER 22 REVIEW

Organic Chemistry

SECTION 2 SHORT

ANSWER Answer the

following questions in

the space provided. \_\_\_\_

1. \_\_\_\_ Hydrocarbons

that contain only single

covalent bonds

between carbon atoms

are called (a) alkanes.

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(c) alkynes. (b)  
alkenes. (d)  
unsaturated.

**CHAPTER 22 REVIEW**  
**Organic Chemistry**

The simplest organic compounds are the hydrocarbons, compounds composed of carbon and hydrogen atoms only. Some hydrocarbons have only single bonds and appear as a chain (which can be a straight chain or can

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have branches) of carbon atoms also bonded to hydrogen atoms. These hydrocarbons are called alkanes (saturated hydrocarbons).

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successful. As  
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does not recommend  
that you have fantastic  
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Organic Chemistry and Hydrocarbons. Review structural formula- p.744. Carbon has 4 valence electrons, thus forms 4 covalent bonds. not only with other elements, but also forms bonds WITH ITSELF....

**chapter25 -  
hydrocarbons -**

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**NAMING** Section Review

HYDROCARBONS Name

Name the compounds below according to the IUPAC naming system

1.  $\begin{array}{cccccccc} & H & H & H & I & I & I & H \\ & | & | & | & | & | & | & | \\ H & - & C & - & C & - & C & - & H \\ & | & | & | & | & | & | & | \\ H & I & I & I & H & H & H & H \end{array}$
2.  $\begin{array}{cccccccc} & H & H & H & H & & & & \\ & | & | & | & | & & & & \\ H & I & I & I & H & - & C & = & C & - & C & - & C \\ & | & | & | & | & & & & & & & & \\ H & C & 4 & H & H & I & a & - & C & - & H & & \\ & & & & & & & & & & & & \\ N & , & 6. & a & - & rn & - & e & - & 4 & pro & p & 5 & t & - \\ n & e & H & CH & 3 & H & I & & & & & & & & \end{array}$

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## **HYDROCARBONS**

**Name - Ms Galloway**

744 Chapter 21 •

Hydrocarbons Section

221.11.1 Introduction

to Hydrocarbons MAIN

Idea Hydrocarbons are

carbon-containing

organic compounds

that provide a source

of energy and raw

materials. Real-World

Reading Link If you

have ridden in a car or

a bus, you have used

hydrocarbons. The

gasoline and diesel fuel

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that are used in cars,  
trucks, and buses

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**Chapter 21:**

**Hydrocarbons**

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11 Chemistry Chapter

13 Hydrocarbons €| ...

22.1 Hydrocarbons

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SECTION 22.1

HYDROCARBONS

(pages 693-701) This section describes the bonding in

hydrocarbons and distinguishes straight-chain from branched-chain alkanes. It also provides rules for naming branch-chained alkanes,

Organic Chemistry and

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Hydrocarbons (pages  
693-694) 1. What

is organic chemistry? It  
the study of the  
chemistry of carbon  
compounds. 2.

**This section  
describes the  
bonding in  
hydrocarbons and ...**

BIGIdeaThe  
substitution of different  
functional groups for  
hydrogen atoms in  
hydrocarbons results in  
a diverse group of

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organic compounds.

22.1 Alkyl Halides and

Aryl Halides MAIN Idea A

halogen atom can

replace a hydrogen

atom in some

hydrocarbons. 22.2

Alcohols, Ethers, and

Amines

**Chapter 22:**

**Substituted**

**Hydrocarbons and**

**Their Reactions**

An editor will review

the submission and

either publish your

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submission or provide  
feedback. Next Answer

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naming organic

chemistry (DOCX 47

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functional groups

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Organic REVIEW packet

(DOCX 76 KB) Review -

ANSWER KEY (PDF 2 ...

## **Unit 11 - Organic Chemistry**

Start studying section 9.1 Carbon

Compounds. Learn vocabulary, terms, and more with flashcards, games, and other study tools.

### **section 9.1 Carbon Compounds**

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## Flashcards | Quizlet

The longest chain containing the double bond has five carbon atoms, so the compound is a pentene (rule 1). To give the first carbon atom of the double bond the lowest number (rule 2), we number from the left, so the compound is a 2-pentene. There is a methyl group on the fourth carbon atom (rule 3), so the compound's name is

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**25.4: Unsaturated  
Hydrocarbons -**

**Chemistry**

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The simplest organic compounds are the hydrocarbons, compounds composed of carbon and hydrogen atoms only. Some hydrocarbons have only single bonds and appear as a chain (which can be a straight chain or can



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have branches) of carbon atoms also bonded to hydrogen atoms. These hydrocarbons are called alkanes (saturated hydrocarbons).

## **1.1. Introduction to organic chemistry | Organic Chemistry**

...

The relative energies of molecular orbitals of given cation are predicted by following

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these steps. Step 1:  
The cyclopropane ring  
is drawn inside the  
circle with its vertices  
touching the circle.  
One of the vertices  
should point down in  
the circle as shown  
below. Step 2 The  
three vertices of ...

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