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elements are periodic functions of their
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chemical properties of the elements are
periodic functions of their atomic
numbers. periodic table. an arrangement
of the elements in order of their atomic
numbers so that elements with similar

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Answers

properties fall in the same column, or group. lanthanides.

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a. 2 and 4 c. 4 and 5 b. 3 and 4 d. 5 and 6
___ 21. The discovery of the noble gases changed Mendeleev's periodic table by adding a new. a. period. c.

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Answers

group. b. series. d. sublevel block. _____

22. In the modern periodic table, elements are ordered according to. a. decreasing atomic mass. c. increasing atomic number. b. Mendeleev's original

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Chapter 5 PRACTICE TEST. Chapter 5

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Answers

PRACTICE TEST Multiple Choice Identify the letter of the choice that best completes the statement or answers the question. _____

_____ 1. The idea of arranging the elements in the periodic table according to their chemical and physical properties is attributed to a. Mendeleev.

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d 6. a 7. b 8. b 9. b 10. a 11. c 12. a 13.
c 14. d 15. c 16. b 17. d 18. a 19. d 20. c

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21. d 22. a 23. a 24. d 25. b

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