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Objectives. After completing this section, you should be able to write the equilibrium constant expression for a given reaction. assess, qualitatively, how far a reaction will proceed in a given direction, given the value of  $K_{eq}$ ; explain the difference between rate and equilibrium.

### **6.7 Describing a Reaction: Equilibria, Rates, and Energy**

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### **6.8: Describing a Reaction - Equilibria, Rates, and Energy**

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