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$H_2O(l) \rightarrow 2NaOH(aq) \cdot K_2O(s) + H_2O(l) \rightarrow 2KOH(aq) \cdot 2K(s) + 2H_2O(l) \rightarrow 2KOH(aq) + H_2(g) + \text{heat energy} \cdot$
 $2Na(s) + 2H_2O(l) \rightarrow 2NaOH(aq) + H_2(g) + \text{heat energy} \dots$

CBSE Class 10th Science Board Exam 2021: Important Chemical Reactions

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